A Convenient Synthesis of 3-Aminotropolone and Its Reactions with Orthoesters. Formation of 8H-Cyclohept[d]oxazol-8-ones

NOTES

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Schmidt reaction of 3-acetyltropolone gave 3-acetamidotropolone, which was hydrolyzed to afford The compound 3 reacted with 3-aminotropolone (3). orthoformate, orthoacetate, orthopropionate, and orthobenzoate to give the corresponding 2-substituted 8H-cyclohept-[d]oxazole-8-one derivatives.

Recently, we found and reviewed that 3-acetyltropolone is very useful as starting material for the synthesis of heterocycle-condensed troponoid compounds.1) In this series, we reported that 3acetyltropolone and its methyl ethers reacted with hydroxylamine to afford 8H-cyclohept[d]isoxazol-8one and 8H-cyclohept[c]isoxazol-8-one as a major and minor product, respectively.2) It is also known that 3aminotropolone reacted with formamide, acetic anhydride, or p-nitrobenzaldehyde to give 8H-cyclo $hept[d]oxazol-8-ones,^{3,4}$ while 2-amino-3-hydroxytropone reacted with acetic anhydride to give 4Hcyclohept[d]oxazol-4-one.^{5,6)} The 1,3-dipolar addition of arenecarbonitrile oxide to tropone gave 4H-cyclohept-[d]isoxazol-4-one derivatives.7)

Previously, 3-aminotropolone was prepared from 3nitrotropolone⁸⁾ or 3-bromotropolone.⁹⁾ On the other hand, it was reported that Schmidt reaction of 4-acetyltropolone afforded 4-aminotropolone.10)

In a series of 3-acetyltropolone chemistry, we wish to report the convenient synthesis of 3-aminotropolone by Schmidt reaction of 3-acetyltropolone and its conversion to 8H-cyclohept[d]oxazole-8-ones by reactions with orthoesters.

Results and Discussion

When we applied the Schmidt reaction to 3-acetyltropolone(1),¹¹⁾ 3-acetamidotropolone(2)³⁾ was isolated. The compound 2 was readily hydrolyzed to give 3-aminotropolone(3), whose averall yield based on 1 was 56%.

A mixture of 3-aminotropolone (3) and triethyl orthoformate or triethyl orthoacetate was heated under reflux to afford respectively 8H-cyclohept[d]oxazole-8one (4a)4) (44%) and its 2-methyl derivative (4b)3) (73%). In the NMR spectrum of the compound 4a, the H-2 proton in the oxazole ring is observed at lower magnetic field (δ =8.21) than the H-2 proton in oxazole

ring $(\delta = 7.95)$.¹²⁾ In a similar manner, the reaction with triethyl orthopropionate and trimethy orthobenzoate gave 2-ethyl- and 2-phenyl-8H-cyclohept[d]oxazol-8-ones (4c and 4d) in 80 and 10% yields, respectively.

Experimental

Measurements. The melting points were determined with a Yanagimoto MP-S2 melting point-measuring apparatus and are uncorrected. The IR spectra were taken on a JASCO IRA-1 spectrophotometer, and the UV spectra on a Hitachi EPS-3T spectrophotometer. The NMR spectra were recorded with a Hitachi R-24 spectrometer (60 MHz).

3-Acetamidotropolone (2). Sodium azide (2.60 g, 40 mmol) was added to a solution of 3-acetyltropolone (1) (3.28 g, 20 mmol) in chloroform (40 ml). Concentrated sulfuric acid (10 ml) was carefully dropped into the mixture in an ice-cooled bath under stirring. After stirring for 2 h at room temperature, the chloroform was removed by decantation. After cold water (80 ml) was added to the residue, a precipitate was collected and recrystallized from methanol to give 2.3 g (66%) of 3-acetamidotropolone (2) as colorless needles: mp 119—120°C (lit,3) 107—108°C); IR (CHCl3) 3315 (NH), 1702 cm^{-1} (C=O); NMR (CDCl₃) δ=2.30 (3H, s, CH₃), 7.0—7.6 (3H, m), 9.1-9.4 (1H, m, NH).

3-Aminotropolone (3). 3-Acetamidotropolone (2) (2.0 g) in 50% sulfuric acid (22 ml) was heated on a water bath for 1 h. The mixture was neutralized with sodium hydrogencarbonate solution, acidified slightly with 30% acetic acid, and extracted with chloroform. The evaporation residue from the extract was recrystallized from benzene-petroleum ether to give 1.3 g (85%) of 3-aminotropolone (3) as yellow needles: mp 86°C (lit,8) 86°C); NMR (CDCl₃) δ =5.6-6.7 (2H, br, NH₂), 6.6—7.5 (m, 4H).

Reaction of 3-Aminotropolone (3) with Triethyl Ortho-A mixture of 3-aminotropolone (3) (137 mg, 1.0 mmol) and triethyl orthoformate (2 ml) was refluxed for 2 h. After cooling, the precipitate was collected and recrystallized from benzene-hexane to give 65 mg (44%) of 8H- $\operatorname{cyclohept}[d]$ oxazol-8-one (4a) as pale yellow needles: mp 149—150°C (lit,4) 150—151°C); IR (CHCl₃) 1625 cm⁻¹ (C=O); UV (CH₃OH) 237 (log ε 4.56), 310 nm (3.94); NMR (CDCl₃) $\delta = 6.8 - 8.0(4H, m), 8.21(1H, s, H-2).$

Reaction of 3 with Triethyl Orthoacetate. A mixture of 3 (274 mg, 2.0 mmol) and triethyl orthoacetate (2 ml) was refluxed for 2 h. The reaction mixture was dissolved in chloroform and twice washed with water. After removal of the solvent, the residue was recrystallized from benzene-hex-

Scheme 2.

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ane to give 234 mg (73%) of 2-methyl-8*H*-cyclohept[*d*]oxazol-8-one (**4b**) as pale yellow needles: mp 150—151 °C (lit,³) 144—145 °C); IR (CHCl₃) 1640 cm⁻¹ (C=O); UV (CH₃OH) 243 (log ε 4.44), 312 nm (3.68); NMR (CDCl₃) δ =2.67 (3H, s, CH₃), 6.9—7.8 (4H, m).

Reaction of 3 with Triethyl Orthopropionate. A mixture of 3 (274 mg, 2.0 mmol) and triethyl orthopropionate (2 ml) was refluxed for 2 h and worked up, as mentioned above, to afford 280 mg (80%) of 2-ethyl-8*H*-cyclohept[*d*]oxazol-8-one (4c) as pale yellow needles: mp 92—93 °C (from benzene-hexane); IR (CHCl₃) 1640 cm⁻¹ (C=O); UV (CH₃OH) 245 (log ε 4.49), 312 nm (3.83); NMR (CDCl₃) δ=1.50 (3H, t, J=7 Hz, CH₃), 3.03 (2H, q, J=7 Hz, CH₂), 6.9—7.8 (4H, m). Found:C, 68.51; H, 5.21; N, 7.92%. Calcd for C₁₀H₉NO₂: C, 68.56; H, 5.18; N, 8.00%.

Reaction of 3 with Trimethyl Orthobenzoate. A solution of 3 (137 mg, 1.0 mmol) and trimethyl orthobenzoate (182 mg, 1.0 mmol) in chloroform (2 ml) was refluxed for 2 h. After removal of the solvent, the residue was chromatographed on a Wakogel B-10 plate $(30\times30~\text{cm}^2)$ with ethyl acetate to give 23 mg (10%) of 2-phenyl-8*H*-cyclohept-[*d*]oxazol-8-one (4d) as pale yellow needles: mp 149—150 °C (from benzene-hexane); IR (CHCl₃) 1640 cm⁻¹ (C=O); UV (CH₃OH) 275 (log ε 4.42), 349 (sh, 3.84), 365 nm (sh, 3.63); NMR (CDCl₃) δ =7.0—7.8 (7H, m,), 8.1—8.5 (2H, m, H-2', 6'). Found: C, 75.32; H, 4.06; N, 6.28%. Calcd for C₁₄H₉NO₂: C,75.50; H, 3.96; N, 6.26%.

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